Acremonol and Acremodiol, New Fungal Bislactones

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In the course of our continuing search for new fungal metabolites with antibacterial properties we isolated new bislactones, acremonol (1) and acremodiol (2) (Fig. 1). In this paper we describe the isolation and structural elucidation of 1 and 2 by ESI-MS, FAB-MS and NMR spectroscopy.

The new bislactones 1 and 2 possess structures related to grahamimycin¹⁾ and the colletodiol family^{2,3)}. For the latter compounds chemical synthesis was reported⁴⁾. Biosynthetic studies⁵⁾ revealed that these compounds are polyketides. The producer strain of 1 and 2 was isolated from a soil sample of the Bermuda Islands as a taxonomically unidentified *Acremonium*-like anamorphic fungus. It forms mycelial colonies colored pale pink to pale brown. The hyphae are mostly narrow, seldom up to $6 \,\mu$ m in diameter. The characteristic conidia are ellipsoid to short cylindrical, mostly $3.5 \sim 6.5/2 \sim 3 \,\mu$ m.

The producer strain was cultivated as surface culture at 24°C in 500 ml Erlenmeyer bottles containing 100 ml medium composed as follows (g/liter): malt extract 20, glucose 10, yeast extract 1, pH 6.0. Each bottle was inoculated with a 1 cm² area of a 15 days agar culture. The cultivation time as stationary culture was three weeks at 24°C. Acremonol (1) and acremodiol (2) were extracted from 40 liters culture broth with 20 liters of ethyl acetate, which was subsequently dried (Na_2SO_4) and evaporated to yield 2.2 g of an oily residue. This was chromatographed on silica gel 60 (Merck, $0.063 \sim 0.1$ mm, column 600×30 mm). Sequential elution occurred with 40 ml portions of CHCl₃, CHCl₃-MeOH (95:5, 80:20, 60:40, 0:100, v/v). Fractions of similar composition, as determined by TLC, were pooled to yield 63 mg of a mixture of 1 and 2. Further purification was carried out by chromatography on Sephadex LH-20

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(column 400×20 mm, eluent MeOH - CHCl₃, 8:2, v/v) and preparative TLC (Merck, silica gel 60 F_{254} , 20×20 cm, eluent CHCl₃-MeOH, 9:1, v/v). Yield: 11 mg of acremonol (1) and 23 mg of acremodiol (2). The physicochemical properties of 1 and 2 are shown in Table 1. The molecular weight and the chemical formula of the new metabolites were determined by HREI-MS (doublefocussing mass spectrometer MAT 95XL, Finnigan, Bremen, Germany) (Table 1) from *m/z* 282.1200 (1; M⁺; calcd. 282.1103) and *m/z* 286.1402 (2; M⁺; calcd. 286.1416). The positive ion ESI-MS (triple quadrupole mass spectrometer Quattro 400, VG Biotech, Altrincham, U.K.) showed *m/z* 283.2 [M+H]⁺ for 1 and *m/z* 287.4 [M+H]⁺ for 2.

The structures of compounds **1** and **2** were determined by one- and two-dimensional NMR spectroscopy (¹H, ¹³C, DEPT, COSY, HMQC, HMBC, Table 2, Bruker Avance DRX 500, Germany).

The positions of substituents and double bonds in the 14membered ring system were assigned due to observable cross peaks in the ¹H, ¹H-COSY spectrum and C, H longrange correlations (HMBC).

The values of coupling constants of neighbouring

Fig 1. Structures of acremonol (1) and acremodiol (2).



	1	2	
Appearance	colorless solid	colorless solid	
Melting point ^a	83 - 84 °C	110 - 112 °C	
Chemical formula	$C_{14}H_{18}O_6$	$C_{14}H_{22}O_{6}$	
HREI-MS m/z	282.1200 [M] ⁺	286.1402 [M] ⁺	
	calcd. 282.1103	calcd. 286.1416	
$[\alpha]_{D}^{25}$ (c 0.3, MeOH)	+ 40 °	+ 98 °	
IR (KBr) cm ⁻¹	1715, 1651, 1648	1690, 1655, 1647	
R_{f} values ^b	0.51	0.38	

Table 1. Physico-chemical properties of acremonol (1) and acremodiol (2).

^a Büchi Melting Point B 540

^b CHCl₃-MeOH (9:1, v/v)

Table 2. ¹H and ¹³C assignments of acremonol (1) and acremodiol (2) (in DMSO- d_6 , chemical shifts in ppm, multiplicity in parentheses (s: singlet, d: doublet, t: triplet, m: multiplet), coupling constants in Hz).

Position	1			2		
	δ _C	δ _H	multiplicity coupling constants	δ _C	δн	multiplicity coupling constants
2	163.9	-	-	174.7	-	-
3	136.7	7.35	d, 16.2	32.0	2.56, 2.86	m, m
4	129.8	6.37	d, 16.2	26.3	1.90, 2.12	m, m
5	200.1	-	-	72.7	3.70	m
6	75.3	5.15	q, 7.2	73.6	5.27	d, q, 6.5, 2.1
8	165.7	-	-	164.9	-	-
9	118.1	5.92	dd, 16.3, 1.7	120.6	6.02	dd, 15.9, 2.2
10	154.3	7.18	dd, 16.3, 3.1	149.2	6.67	dd, 15.9, 2.6
11	68.8	4.35	m	69.2	4.60	m
12	30.4	1.48, 1.87	m, m	28.2	1.78	m
13	27.6	1.66, 1.55	m, m	26.9	1.52, 1.82	m, m
14	71.9	4.89	m	70.6	4.95	m
15	15.8	1.48	d, 7.2	16.7	1.24	d, 6.4
16	18.5	1.20	t, 7.0	19.2	1.16	d, 6.6

olefinic protons (Table 2) confirmed *E*-configuration for the double bonds. The absence of NOE effects between H-11/H-16 and H-15/H-16, respectively, in the NOESY spectra was compatible with the view that relative

stereochemistry of 1 and 2 is the same as was reported for colletodiol^{2,3)}. The ${}^{3}J_{\text{H-5,H-6}}=2.1 \text{ Hz}$ suggested synclinal proton conformation and α -positions of substituents at C-5 and C-6.

Microorganisms	Diameter of inhibition zone (mm)			
	1	2		
Bacillus subtilis ATCC 6633	24	22		
Sporobolomyces salmonicolor	0	0		
Penicillium notatum JP 36	18 p ^a	16 p		
Escherichia coli SG 458	14 p	13 p		
Enterococcus faecalis	21	16		
Staphylococcus aureus	26	20		
Pseudomonas aeruginosa	13 p	12 p		

Table 3. Antimicrobial activity of acremonol (1) and acremodiol (2).

Each well contained 50 µg of 1 or 2 dissolved in 50 µl methanol

^a p = partial inhibition

The new metabolites 1 and 2 display antimicrobial activity against a series of Gram-positive bacteria and fungi in the agar well diffusion $assay^{6,7}$ (Table 3). In contrast, colletodiol described as biologically inactive.

Moreover, 1 and 2 were active in cellular phagocytosis assay using dog PMNL cells⁸⁾ decreasing the oxidative burst at concentrations $\ge 4 \,\mu g/ml$.

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